# Lochromolide: A New Acetylated Withanolide from lochroma coccineum 

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# IOCHROMOLIDE: A NEW ACETYLATED WITHANOLIDE FROM IOCHROMA COCCINEUM 

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#### Abstract

The phytochemical investigation of Iocbroma cocineum (Solanaceae) afforded three withanolides: withaferine A [1], withacnistine [2], and a new compound, the (17R,20S,22R)$4 \beta$-hydroxy-1-oxo-5 $\beta, 6 \beta$-epoxy-16 $\alpha$-acetoxywitha- 2,24 -dienolide, which we called iochromolide [3]. The structure of $\mathbf{3}$ was deduced on the basis of nmr and mass spectral comparison with $\mathbf{1}$ and 2 and definitively established by X-ray crystallography.


Withanolides are natural steroidal lactones produced as far as known only by plants in the Solanaceae ( 1,2 ). They can be classified into four main groups (1): 20deoxywithanolides (withaferine A and related compounds), 20-hydroxywithanolides (e.g., withanolide D), withanolides with the C-17 side chain $\alpha$-oriented (like withanolide E), and withanolides whose basic framework is modified or chlorinated (physalins, withaphysalins, physalolactones, etc.). It has been observed that one chemotype or one given species generally produces withanolides of one of these types only. The biological and pharmacological properties of withanolides are of great interest. Although only a few of them have been tested, we know that they may exhibit antitumor, antifeedant, immunomodulating, and anti-inflammatory activities (1,2). During our investigation of Iochroma coccineum Scheidweiler (3) we isolated three substances belonging to the 20 -deoxywithanolide class. The chemistry of the genus Iochroma Benth. (4) has, to our knowledge, not been studied before, except for the recent isolation of three hydroxycinnamic acid amides from Iochroma cyaneum (5).

## RESULTS AND DISCUSSION

The defatted $\mathrm{H}_{2} \mathrm{O} / \mathrm{MeOH}$ extract of the leaves of $I$. cocineum furnished, after extrac-

tion with $\mathrm{Et}_{2} \mathrm{O}$ and separation by various chromatographic steps, three withanolides (see Experimental).

The ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-nmr spectra (Tables 1 and 2), the mass spectra, the uv absorption maxima, and the melting points allowed us to identify two of them as withaferine A [1] and withacnistine [2] by comparison with earlier reported data ( $6-10$ ).

The uv maximum of the third isolated compound 3, taken in MeOH , is 216 nm with $\log \epsilon 4.16$, which is typical for withanolides, since both unsaturated lactones and enone groups (ring A) exhibit absorption maxima between 215 and 230 nm with high molar extinction coefficients $(11,12)$.

The mass spectrum of 3 showed a molecular peak at $m / z 512$. The ion at $m / z 452$, corresponding to the loss of HOAc, indicated the presence of an acetate group. The presence of these two peaks in the mass spectrum of 2 let us assume the new compound to be an isomer of 2. Fission between $\mathrm{C}-20$ and $\mathrm{C}-22$ led to the base peak at $m / z 125$ $\left(\mathrm{C}_{7} \mathrm{O}_{2} \mathrm{H}_{9}\right)$, suggesting the absence of an OH group on the side chain, since hydroxylated lactones, like $\mathbf{1}$, give an intense ion at $m / z 141$, corresponding to $\mathrm{C}_{7} \mathrm{O}_{3} \mathrm{H}_{9}(6)$. The ion corresponding to the loss of the side chain including two carbons of ring D (C-16 and C-17), present in the spectra of $\mathbf{1}(\mathrm{m} / \mathrm{z} \mathrm{197)}$ ) and $\mathbf{2}(\mathrm{m} / \mathrm{z} 181)$, was missing in the spectrum of 3; this could be an indication for a change of structure occurring at ring D. The major fragmentation ions showed further fissions corresponding to the loss of $\mathrm{H}_{2} \mathrm{O}$ and HOAc: in addition to $m / z 452$ we could observe $m / z 494$ ( $512-18$ ), $m / z 434$ ( $512-60-18$ ), $m / z 281(299-18)$. Other typical ions were present ( $m / z 55,67,79$, $97,105,267,285,299$ ), resulting from the cleavage of the steroidal skeleton, as well as the ion $m / z 124$ due to the fission of ring B with possible hydrogen transfer; all these peaks were observed in the mass spectra of the three isolated withanolides.

Table 1. ${ }^{1} \mathrm{H}-\mathrm{nmr}$ Spectral Data of Relevant Protons of 1-3. ${ }^{\text {a }}$

| Proton | Compound |  |  |
| :---: | :---: | :---: | :---: |
|  | 1 | 2 | 3 |
| H-2 | 6.21 (d, 10.1) | 6.20 (d, 10.1) | 6.21 (d, 10.1) |
| H-3 | 6.94 (dd, 6.1, 10.1) | 6.94 (dd, 5.8, 10.1) | 6.95 (dd, 5.8, 10.1) |
| H-4 | 3.77 (d, 6.1) | 3.77 (d, 5.8) | 3.77 (d, 5.8) |
| H-6 | 3.24 (br s) | 3.25 (br s) | 3.24 (br s) |
| H-7b | 2.16 (dt, 3.2, 5.0, 14.9) | 2.16 (dt, 2.9, 4.8, 14.9) |  |
| H-16 | - | - | 4.90 (t, 7.2, 9.0) |
| H-18 | 0.72 (s) | 3.83 (d, 11.9) ${ }^{\text {c }}$ | 0.76 (s) |
|  |  | 4.21 (d, 11.9) |  |
| H-19 | 1.42 (s) | 1.41 (s) | 1.41(s) |
| H-21 | 0.99 (d, 6.6) | 1.12 (d, 6.5) | 1.02 (d, 6.5) |
| H-22 | 4.43 (dt, 3.2, 4.3, 13.3) | 4.38 (dt, 3.2, 4.3, 13.3) | 4.17 (dt, 3.2, 4.3, 13.1) |
| H-23b | 2.50 (dd, 13.3, 18.0) | 2.45 (brt) | 2.42 (br t) |
| H-27 | $\begin{aligned} & 4.35(\mathrm{~d}, 13.0)^{\mathrm{c}} \\ & 4.39(\mathrm{~d})^{\mathrm{d}} \end{aligned}$ | 1.88 (s) | 1.88 (s) |
| H-28 | 2.04 (s) | 1.94 (s) ${ }^{\text {e }}$ | 1.93 (s) ${ }^{\text {c }}$ |
| H-30 ${ }^{\text {f }}$ | - | 2.08 (s) ${ }^{\text {e }}$ | 1.96 (s) ${ }^{\text {e }}$ |

${ }^{2}$ Chemical shifts are reported in ppm , signal multiplicities and coupling constants $(\mathrm{Hz})$ are shown in parentheses.
${ }^{\text {b }}$ Signal overlapped.
${ }^{\text {c }} \mathrm{AB}$ system.
${ }^{\mathrm{d}} J$ not measurable, the signal being overlapped by that of $\mathrm{H}-22$.
${ }^{\text {e }}$ Signals within a vertical column may be interchanged.
${ }^{\text {f }}$ Protons from -OAc.

Table 2. ${ }^{13} \mathrm{C}$-nmr Spectral Data of 1-3. ${ }^{2}$


[^0]${ }^{13} \mathrm{C}$-nmr data of $\mathbf{3}$ (Table 2) were quite similar to those of 2 : both contained signals for 30 carbons, but the DEPT analysis indicated the presence of one CH and one Me more and two $\mathrm{CH}_{2}$ less for $\mathbf{3}$ than for $\mathbf{2}$. The triplet assigned to $\mathrm{C}-18$ in $\mathbf{2}$ disappeared, but one additional quartet was observed at 13.0 ppm which corresponds to the signal for $\mathrm{C}-18$ ( Me ) of $\mathbf{1}$. On the other hand, the triplets at about $\delta 24$ and $\delta 27$, due to $\mathrm{C}-15$ and C-16 and present in the spectra of $\mathbf{1}$ and $\mathbf{2}$, were lacking, but the spectrum of 3 showed another triplet at $\delta 34.6$ and a further doublet at $\delta 79.5$. The environment of C 15 and $\mathrm{C}-16$, i.e., $\mathrm{C}-13, \mathrm{C}-14$, and $\mathrm{C}-17$, underwent small changes in chemical shifts. The signals for all the other carbons showed similar multiplicities and shifts. Therefore, we expected the third withanolide to be isomeric to 2 , with an acetoxyl group on C-15 or $\mathrm{C}-16$ and with $\mathrm{C}-18$ being a methyl.

The H -nmr spectrum of $\mathbf{3}$ (Table 1) also showed strong similarities with those of $\mathbf{1}$ and 2 but without the $A B$ system ( $\mathrm{H}-18$ ) of 2 ; six methyl groups were present, giving five singlets corresponding to $\mathrm{Me}-18$ (acetoxylated in 2), Me-19, Me-27 (hydroxylated in 1), Me-28 and Me-30 (Me from aceroxyl), and one doublet for Me-21, as well as one hydroxyl group ( $\delta 2.59$ ) which proton was exchangeable with $\mathrm{D}_{2} \mathrm{O}$. The spectrum was


Figure 1. 2D ${ }^{1} \mathrm{H}_{-}^{13} \mathrm{C}$ heteronuclear shift correlation of iochromolide [3].
in good agreement with the presumed structure but did not indicate whether the acetoxyl group was on $\mathrm{C}-15$ or $\mathrm{C}-16$. Figure 1 shows the ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ heternuclear shift correlation spectrum of 3 .

The definitive structure of $\mathbf{3}$ was determined by $\mathbf{X}$-ray crystallography (Figure 2), which confirmed the above assumprions and clearly indicated that the acetoxyl was attached at C-16 and $\alpha$-oriented. This compound constitutes the first known example of a 16-acetoxywithanolide. No essential differences were observed between the two independent molecules of the asymmetric unit. The relative stereochemistry being fixed by the X-ray structure determination, the absolute configuration of the molecule has been


Figiliti: 2. Stereoscopic view of iechromolide [3].
assumed as equivalent to those of withaferin A derivatives (13). An analysis of the ring conformations of the steroid is reported in Table 3. The ring fusions ( $A / B$ cis; $B / C$ trans; $\mathrm{C} / \mathrm{D}$ trans; 9,10 anti; 8,14 anti) and geometrical parameters of both molecules are close to those observed for withaferin A. The molecular packing is fixed by hydrogen bonds involving the $4 \beta$-hydroxyl and the $16 \alpha$-acetoxyl groups: $O(5) \ldots(02)_{x+1, y+1, z-1}=$ $2.787(13) \AA, \mathrm{O}(2) \ldots \mathrm{O}(05)_{\mathrm{x}, \mathrm{y}+1, z-1}=2,800(13) \AA$.

## EXPERIMENTAL

General experimental procedures.-Melting points were measured on a Tortoli apparatus and are uncorrected. Uv spectra were determined on a Lambda 3 uv/vis spectrophotometer (Perkin-Elmer). ${ }^{1} \mathrm{H}$ - and ${ }^{14} \mathrm{C}$-nmr spectra were recorded on a Bruker AMX 360 at 360.13 and 90.53 MHz , respectively, in $\mathrm{CDCl}_{3}$ at $23^{\circ}$, using TMS as internal standard ( $\delta$ ). Eims were determined at 70 eV on a VG 7070 E spectrometer. Analytical and semi-preparative hplc were performed on a $W$ aters 600 E apparatus, using respectively a Waters 990 Photodiode Array Detector and a Waters 484 Tunable Absorbance Detector.

Plant material.-Seeds of $I$. cocineum (3) were provided by the Jardim Botanico da Universidade de Lisboa, Portugal. Plants were grown at the Station Fédérale de Recherches Agronomiques, Centre des Fougères, in Conthey, Wallis, Switzerland. A voucher specimen is deposited in our laboratory. Leaves were dried at $40^{\circ}$.

Extraction and isolation of 1, 2, and 3.-Powdered dried leaves of 1 . cocineum ( 100 g ) were extracted with $\mathrm{MeOH} 80 \%$ at $40^{\circ}$ for 15 h . After filtration and evaporation, the residue was dissolved in

|  | Mol. 1 | Mol. 2 |
| :---: | :---: | :---: |
| Ring A | (C-1-C-2-C | 4-C-5-C-10) |
| Conformation | twist-boat | twist-boat |
| $\mathrm{Q}_{2}$ | 0.4613 | 0.3918 |
| $\mathrm{Q}_{3}$ | 0.1671 | 0.1686 |
| $\mathrm{Q}_{\mathrm{T}}$ | 0.4907 | 0.4265 |
| $\varphi_{2}$ | -148.2 | -156.5 |
| $\Theta_{2}$ | 70.1 | 66.7 |
| $\Delta \mathrm{C}_{2}$ | C-1-C-2: 0.006 (7) | C-1-C-2: 0.023 (6) |
| Ring B | (C-5-C-6-C | 8-C-9-C-10) |
| Conformation | half-chair | half-chair |
| $\mathrm{Q}_{2}$ | 0.3753 | 0.3690 |
| Q ${ }_{3}$ | 0.3541 | 0.2963 |
| $\mathrm{Q}_{\text {T }}$ | 0.5160 | 0.4732 |
| $\varphi_{2}$ | -165.8 | -166.0 |
| $\Theta_{2}$ | 46.7 | 51.2 |
| $\Delta \mathrm{C}_{2}$ | C-5-C-6:0.057(5) | C-5-C-6: 0.057 (5) |
| Ring C | (C-8-C-9-C-1 | 2-C-13-C-14) |
| Conformation | chair | chair |
| $\mathrm{Q}_{2}$ | 0.0668 | 0.0932 |
| $\mathrm{Q}_{3}$ | 0.5641 | 0.5508 |
| $\mathrm{Q}_{\text {T }}$ | 0.5681 | 0.5587 |
| $\varphi_{2}$ | -108.1 | -107.8 |
| $\Theta_{2}$ | 6.8 | 9.6 |
| $\Delta \mathrm{C}_{2}$ | C-9-C-11:0.013(5) | C-9-C-11:0.017(5) |
| Ring D | (C-13-C-14 | C-16-C-17) |
| Conformation | envelope | envelope |
| $\mathrm{Q}_{2}$ | 0.4410 | 0.4401 |
| $\varphi_{2}$ | -173.5 | -174.94 |
| $\Delta \mathrm{C}_{5}$ | C-13:0.043 (6) | C-13:0.033 (6) |

$\mathrm{MeOH}-\mathrm{CCl}_{4}-\mathrm{H}_{2} \mathrm{O}$ ( $57: 20: 18$ ); after separation from the $\mathrm{CCl}_{4}$ layer, the $\mathrm{H}_{2} \mathrm{O} / \mathrm{MeOH}$ layer was further extracted with $\mathrm{CCl}_{4}$ and, after elimination of MeOH , with $\mathrm{Et}_{2} \mathrm{O}$. The $\mathrm{Et}_{2} \mathrm{O}$ phase gave, after evaporation, 700 mg of residue which was chromatographed over a Sephadex LH-20 (Pharmacia) column ( $2 \times 60 \mathrm{~cm}$ ) with a gradient of MeOH in $\mathrm{H}_{2} \mathrm{O}$ ( 0 to $100 \%$ of $\mathrm{MeOH}, 10 \%$ increments). The fractions corresponding to 20 to $60 \%$ of MeOH were collected, the residue obtained after evaporation of the solvent was applied on a Lobar Lichroprep Si 60 column (A size, Merck), and the elution was performed with hexane/ $\mathrm{CHCl}_{3}$ (gradient from 20 to $100 \%$ of $\mathrm{CHCl}_{3}, 5 \%$ increments). An impure mixture ( 90 mg ) of $\mathbf{2}$ and $\mathbf{3}$ was obtained with 25 to $40 \%$ of $\mathrm{CHCl}_{3}$, and $\mathbf{1}(27 \mathrm{mg})$ was eluted with 50 to $55 \%$ of $\mathrm{CHCl}_{3}$. Compounds $\mathbf{2}$ and $\mathbf{3}$ were isolated by semi-preparative hplc $\left[\mu\right.$ Bondapak $\mathrm{C} 18,19 \times 150 \mathrm{~mm}, \mathrm{MeCN}-\mathrm{H}_{2} \mathrm{O}(40: 60), 10 \mathrm{ml} \cdot \mathrm{min}{ }^{1}$, detector at 215 nm . Compound $\mathbf{3}$ was crystallized from ErOAc and gave white needles. Compound $\mathbf{1}$ was further purified by crystallization from $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}$ and $\mathrm{Me} 2 \mathrm{CO} /$ hexane.

Fractions were monitored by analytical hplc [ $\mu$ Bondapak $\mathrm{C} 18,3.9 \times 30 \mathrm{~mm}, \mathrm{MeCN}-\mathrm{H}_{2} \mathrm{O}(40: 60)$, $1.3 \mathrm{ml} \cdot \mathrm{min}^{-1}$, detection at 215 and 227 nm$]$ and by tk. The methods used for thc were: (a) (original method) i -BuCOMe-HexOH-hexane-HOAc (30:30:40:1) saturated with $\mathrm{H}_{2} \mathrm{O}$, on Si gel $60 \mathrm{~F}_{254}$ plates (Merck), $\mathbf{1} R_{f} 0.27,2 R_{f} 0.47,3 R_{f} 0.52$, and (b) MeOH- $\mathrm{H}_{2} \mathrm{O}(70: 30)$, on $\mathrm{RP} 18 \mathrm{~F}_{254} 5$ plates (Merck), $\mathbf{1}$ $R_{f} 0.28,2 R_{j} 0.23, \mathbf{3} R_{f} 0.23$. Compounds 1, $\mathbf{2}$, and $\mathbf{3}$ were positive with Dragendorff's reagent and also gave positive reactions for epoxide (14).

Crystallographic data and structure refinement of iochromolide [3]- $\mathrm{C}_{30} \mathrm{H}_{40} \mathrm{O}_{7}$ / $\mathrm{C}_{4} \mathrm{H}_{8} \mathrm{O}_{2}, \mathrm{~mol} \mathrm{wt}=512 . / 88$, crystallized in the triclinic system, space group $P l$, with two independent molecules in the asymmetric unit and one molecule of ErOAc used as solvent of crystallization; $a=9.43 \mathrm{~s}$ (2), $b=13.603$ (3),$c=14.468$ (4) $\AA, \alpha=65.26(2), \beta=70.98(2), \gamma=69.67(2)^{\circ}, Z=2, d_{c}=1.20 \mathrm{~g} \cdot \mathrm{~cm}^{\text { }}$, $\mu=0.079 \mathrm{~mm}^{-1}, F o o o=600$. The diffracted intensities were measured at room temperature on a Nonius CAD4 diffractometer with graphite monochromated $\mathrm{MoK} \alpha$ radiation. Data collection: $\sin \theta / \lambda<0.53$; $b=-9,9 ; k=-14,14 ; t=0,15, \omega / 2 \theta$ scans, no absorption correction. Of the 3844 measured reflections, 3086 were considered as observed $[\mid$ Fol $>4 \sigma($ Fo $)$ ] and used in structure refinement. The structure was solved by direct methods (MULTAN-87) (15) and refined by full-matrix least-squares analysis. Atomic scattering factors and anomalous-dispersion terms are taken from Internatiunal Tables for X -ray Crystallography (16). All coordinates of the H atoms were calculated. The hydrogen atom of the hydroxyl group has nor been observed. The ErOAc molecule is disordered, and seven aromic sites have been observed and refined with fixed values of population parameters (0.5) and isotropic displacement parameters ( $\mathrm{U}=0.05 \AA^{2}$ ). The maximum and minimum residual electron densities in the final $\Delta \mathrm{F}$ synthesis were 0.59 and $-0.35 \mathrm{e} \AA^{-3}$, respectively. The final $R$ factor was $R=0.087\left[\omega R=0.062, w=1 / \sigma^{2}(\mathrm{Fo})\right]$. All calculations were performed with the XTAL-3.0(17) program. ${ }^{\prime}$

Withaferine A [1]. - Mp after crystallization from $\mathbf{M e O H} / \mathrm{H}_{2} \mathrm{O} \quad 230-235^{\circ}$, from $\mathrm{Me}_{2} \mathrm{CO} /$ hexane $247-248^{\circ}$; eims $\mathrm{m} / \mathrm{z}(\%)[\mathrm{M}]^{+} 470(11),\left[\mathrm{M}-\mathrm{H}_{2} \mathrm{O}\right]^{+} 452(6),\left[452-\mathrm{H}_{2} \mathrm{O}\right]^{+} 434(4),\left[434-\mathrm{H}_{2} \mathrm{O}\right]^{+}$ 416 (3), 387 (13), 347 (24), 329 (5), 311 (5), 285 (22), 267 (20), 241 (34), $197(21), 141(81), 124(100)$, 123 (58), 105 (35), 95 (89), 79 (31), 67 (58), 55 ( 58 ); uv $\lambda \max (M e O H) 214$ and $333 \mathrm{~nm} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ see Table $1 ;{ }^{13} \mathrm{C}$ nmr see Table 2.

Withacnistine $[\mathbf{2}]$. -Mp after crystallization from $\mathrm{C}_{6} \mathrm{H}_{6} 136-142^{\circ}$; eims $m / z(\%)[\mathrm{M}]^{+} 512$ (3), $\left[\mathrm{M}-\mathrm{H}_{2} \mathrm{O}\right]^{+} 494(3), 466(3),[\mathrm{M}-\mathrm{HOAc}]^{+} 452(4), 434(6), 406(7), 389(11), 329$ (4), 311 (5), 285 (8), 181 (10), 157 (10), 145 (12), 154 (11), 125 (100), 124 (20), 105 (12), 97 (21), 79 (10), 67 (15), 55 (12); uv $\lambda \max (\mathrm{EtOH} 95 \%) 216$ and $330 \mathrm{~nm} ;{ }^{1} \mathrm{H}$ nmr see Table $1 ;{ }^{1} \mathrm{C}$ nmr see Table 2.
lachromolide [3].-Mp after crystallization from EtOAc 146-154 ${ }^{\circ}$; eims $m / z(\%)[M]{ }^{+} 512(0.7)$, $\left[\mathrm{M}-\mathrm{H}_{2} \mathrm{O}\right]^{+} 494(0.6),[\mathrm{M}-\mathrm{HOAc}]^{+} 452(2), 434(6), 419(2), 329(2), 299(7), 285(5), 281(7), 267$ (2), 175 (9), 152 (12), 125 (100), 124 (12), 105 (9), 97 (18), 79 (9), 67 (11), 55 (10); uv $\lambda \max (\mathrm{MeOH})$ 216 and 330 nm ; ${ }^{1} \mathrm{H}$ nmr see Table $1 ;{ }^{17} \mathrm{C}$ nmr see Table 2.

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[^1]
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[^0]:    ${ }^{\mathrm{a}}$ Chemical shifts are reported in ppm ; signal multiplicities are shown in parentheses.
    ${ }^{\mathrm{b}}$ Signals within a vertical column may be interchanged.

[^1]:    ${ }^{1}$ Atomic coordinates for this structure have been deposited with the Cambridge Crystallographic Data Centre and can be obtained on request from Dr. Olga Kennard, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, UK.

